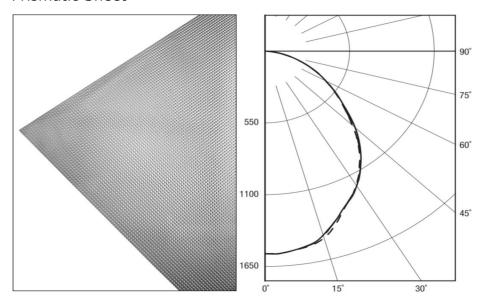
### **Ingemann Components**

### MicroMid<sup>™</sup>

**Prismatic Sheet** 



#### **Description:**

MicroMid<sup>™</sup> Sheet features 1.5mm wide male pyramid prisms to achieve high angle brightness control, excellent lamp obscuration and high transmittance. MicroMid is made from PMMA or PC in 2mm and 3mm thickness and frost acrylic in 2mm thickness upon request.

#### **Application:**

The MicroMid is intended for applications in which high angle brightness cannot be tolerated.

#### **Service information:**

For samples, pricing and delivery please contact us at: +45 4618 6644 Email:

sales@ingemanncomponents.com

Looking for a solution with this product, click <u>here</u>.

#### **Product data** Standard Material PMMA - clear acrylic PC - clear polycarbonate (Frost acrylic upon request) Available size Square 1270mm x 1270mm Thickness 2mm ± 0.127mm 3mm +0.127 to +0.254mm Pyramid width 1.5 mm **Refractive Index** 1.491 for PMMA 1.585 for PC **Transmittance** 90% -40°C to +80°C **Temperature Range**

### MICROMID<sup>TM</sup>

High transparent microstructured pyramids that provides the excellent de-glaring effect and high transmittance.

90% Transmission 2-3mm thickness Max temp +80°C Custom sizes available

Ingemann Components Tingbjergvej 6 4632 Bjæverskov Denmark





# **Ingemann Components**

# Technical Specs - MicroMid™

Properties	3 mm PMMA   PC	Notes
Physical -	•	
Density	PMMA: $1.18 \frac{g}{cm^3}$	
	PC: $1.2 \frac{\ddot{g}^{m}}{cm^{3}}$	
Rockwell Hardness	PMMA: 113	
	PC: 108	
Optical -		
Transmittance	90%	
Refractive index	PMMA: 1.491	
	PC: 1.585	
Reflection	N/A	
Mechanical -		
Tensile strength	PMMA: 69.9 <i>MPa</i>	
	PC: 70 <i>MPa</i>	
Thermal -		
Long term temp.	PMMA: -40°C to 80° PC: -40°C to 120	
Short term temp.	PMMA: 95°C	
	PC: 130°C	
Melting temp.	PMMA: 130°C PC: 288°C	
Surface	Prismatic pyramids other side	on one side, glossy on
UV stable	Yes	
Dirt depreciation	Anti-static treatment	
Chemical Resistance		See next page
Thermal expansion	PMMA: $7 \cdot 10^{-5} / K$	Also used units
	PC: $6.5 \cdot 10^{-5} / K$	$K^{-1}x10^{-5}$ or $10^{-6}/K$
Glow wire test	PC GWFI: 960/2	960°C
IEC 60695-2-12	Results: Pass	2 mm thickness
Fire Rating	PMMA	Class B2 (DIN 4102)

# **Processing options at Ingemann Components**

Processing	Yes/No	Notes	
Milling	Yes	Recommended processing	
CNC Knife	No		
Laser Cutting	No		
Saw	Yes		
Die Cut	No		
Thermo-forming	No		
Print	Yes		

 $\mathsf{MICROMID}^\mathsf{TM}$ 

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### **Ingemann Components**

### **Chemical Resistances for PMMA**

Chemical resistance at 20°C			
Acetone	-	Ethyl acetate	-
Ammonia	+	Glycerin	+
Amyl Alcohol	-	Fuel oil	0
Benzene, free from aromatics	-	Hexane	+
Benzole	-	Isopropanol	0
Boric Acid	+	Coffee	+
Butanol	-	Caustic potash solution	+
Chlorinated hydro-carbon	-	Ketone	-
Chloroform	-	Methylene chloride	-
Chlorinated water/air	0	Lactic acid 10%	+
Dibutyl phthalate	-	Mineral oil	+
Dioctyl phthalate	-	Caustic soda	+
Glacial acetic acid	-	Nitrocellulose lacquer	-
Acetic essence	-	Oxalic acid	+
Aqueous acetic acid	+	Wax	+
Ethanol	0	Hydrogen peroxide	0
Acidity of wine	+	Hydrochloric acid conc. 35%	+
Xylene	-	Sodium carbonate	+
Paraffin	+	Salad vinegar	+
Petroleum ether	+	Stearic Acid	+
Phosphoric acid 10%	+	Tea	+
Sulphuric acid 10%	+	Turpentine	+
Nitric acid 10%	+	Toluene	-
Hydrochloric acid 10%	+	Diluting agent	-

- + Resistant
- o Limited resistance
- Not Resistant
- na Not available

# $MICROMID^{TM}$

At 20°C PMMA is resistant to hydrocarbons, aromatic free carburetor fuel, mineral oils, vegetable-and animal fats and oils, water, aqueous salt solutions, diluted acids and alkalis. Aromatic hydrocarbons and hydrogen chlorides, ester, ether and ketones attack and degrade PMMA.

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# Ingemann Components Chemical Resistances for PC

-	Ethyl acetate	-
1	Glycerin	+
1	Fuel oil	+
-	Hexane	+
-	Isopropanol	+
+	Coffee	+
+	Caustic potash solution	-
-	Ketone	-
-	Methylene chloride	-
0	Lactic acid 10%	+
-	Mineral oil	+
-	Caustic soda	-
-	Nitrocellulose lacquer	-
na	Oxalic acid	+
+	Wax	na
+	Hydrogen peroxide	+
na	Hydrochloric acid conc. 35%	-
-	Sodium carbonate	+
+	Salad vinegar	na
0	Stearic Acid	+
+	Tea	+
+	Turpentine	-
+	Toluene	-
+	Diluting agent	na
	+ + - 0 - na + + na - + + + + + + +	- Glycerin - Fuel oil - Hexane - Isopropanol + Coffee + Caustic potash solution - Ketone - Methylene chloride 0 Lactic acid 10% - Mineral oil - Caustic soda - Nitrocellulose lacquer na Oxalic acid + Wax + Hydrogen peroxide na Hydrochloric acid conc. 35% - Sodium carbonate + Salad vinegar 0 Stearic Acid + Tea + Turpentine + Toluene

- + Resistant
- o Limited resistance
- Not Resistant
- na Not available

### $MICROMID^{TM}$

Polycarbonate has a good chemical resistance, at room temperature, to a variety of organic and inorganic acids. Water, vegetable oils, solutions of neutral salts, aliphatic hydrocarbons and alcohols.

When polycarbonate is attacked it takes three forms, the first is crystallization which makes the surface white and swelling. This happens with aldehydes, ethers, ketones and aromatic hydrocarbons.

The next is complete destruction, this is caused by alkalines, alkali salts amines and high ozone concentrations.

The third is cracking or crazing to the material, it cracks when stress and acetone or xylene is combined.

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